

Synthetic approach to condensed heterocyclic analogues from etoposide revisited. Synthesis of A-ring pyridazine etoposide

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Abstract—The synthetic approach to condensed heterocyclic analogues from etoposide was revisited. The described procedure allows the synthesis of A-ring pyridazine etoposide **13**, featuring the use of tetrakis(triphenylphosphine)palladium as catalyst under optimized conditions in the key Stille cross coupling between bistriflate **14** and the vinylstannane without epimerization at C-2. The TBDMS-protecting group was critical to cleanly obtain the pivotal intermediate **19**.

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Etoposide (VP-16, Vepesid[®]) **1** is a cancer chemotherapeutic agent¹ widely used in the treatment of small cell lung cancer (SCLC), testicular carcinoma, lymphoma and Kaposi's sarcoma. Topoisomerase II is the target for **1** that induces cell death by enhancing enzyme-mediated double-strand DNA scission.² To overcome drug resistance and poor water solubility observed in clinic with **1**, intensive research has been devoted to the synthesis of etoposide analogues³ (Fig. 1).

In the absence of the 3D structure of the topoisomerase II active site for conducting rational drug design in this

family, a composite pharmacophore model has been proposed by MacDonald et al.^{4,5} for expression of topoisomerase II activity, defining three structural domains: the intercalation-like domain in ternary complex (i.e., the planar polycyclic region), the variable substituent domain (i.e., the C₄ molecular region) and the minor groove binding domain (i.e., the pendant E-ring region). The comparative molecular force field analysis (CoMFA), applied by Lee et al.⁶ to develop QSAR models for epipodophyllotoxins, further specified by the nature of the C₄ molecular region for topoisomerase II inhibition (i.e., bulky substituents).

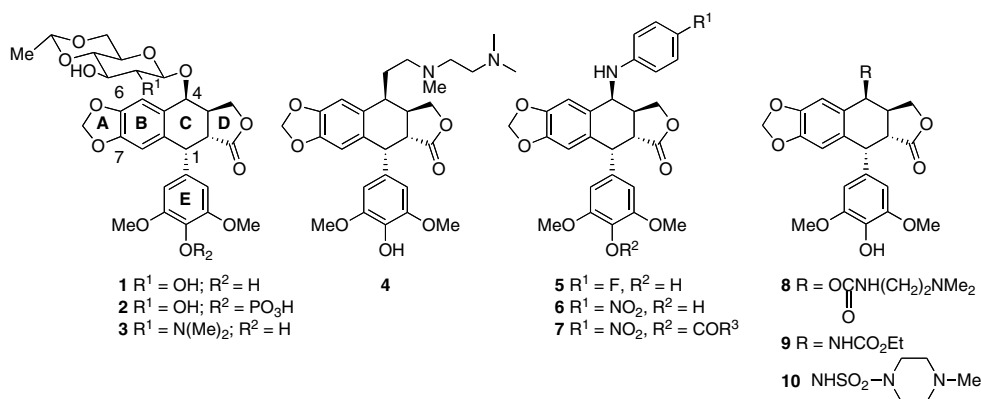


Figure 1. Structures of etoposide **1** and analogues **2–10**.

Keywords: Anticancer drugs; Etoposide; Topoisomerase II; Stille reaction.

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More recently renewed interest in drugs belonging to this class has led to the clinical development of etoposide (Bristol Myers), a 4'-phosphate prodrug of etoposide **2**.⁷ Several non-glycoside derivatives including NK611 (Nippon-Kayaku) **3**, a 2''-dimethylamino-etoposide,⁸ TOP-53 (Taiho) **4**,⁹ NPF **5**,¹⁰ and GL-331 **6**¹¹ and its analogues **7**¹² emerged as promising drug candidates for cancer but led to some unsuccessful phase II clinical trials. However, new developments of these compounds are not excluded. Our continuous interest in the synthesis of etoposide analogues prompted us to extend the evaluation of the variable substituent domain.^{4,5} Thus, incorporation of a 4-*O*- β -carbamate¹³ (**8**) or a 4-*N*- β -carbamate¹⁴ (**9**), or a 4-sulfonamide¹⁵ (**10**) resulted in an increase in the cytotoxicity compared to etoposide **1**. SAR studies centered on A-ring modification of etoposide or non-glycoside analogues turn out to have been far less investigated. Kadow et al.¹⁶ reported the synthesis and in vivo anti-P388 leukemia activity of compounds **11** wherein the 6,7-dimethoxy ($R^1 = R^2 = \text{Me}$) and 6,7-diacetoxy ($R^1 = R^2 = \text{COMe}$) analogues retained the most activity but inferior to etoposide (Fig. 2). These developments emphasize the importance of the methylenedioxy group of **1** for optimal antitumor activity. Lee et al.,¹⁷ also investigated the DNA

intercalating domain of the pharmacophore model by introducing various phenazine rings into 4'-*O*-demethyl-etoposide analogues. Among these compounds, 4'-*O*-demethyl-4 β -(4'''-nitroanilino)-4-desoxy-podophenazine **12** displays an activity superior to etoposide when evaluated against KB and KB/7d (VP-16-Resistant cells) in vitro. It is interesting to note that extension of the phenazine ring to the benzophenazine ring led to a loss of activity. Unlike etoposide, podophenazine **12** was found to be a weak topoisomerase II inhibitor in vitro, but exhibits a novel mechanism of action.

Based on MacDonald's composite pharmacophore model, we designed condensed heterocyclic analogues from etoposide¹⁸ with the aim to enhance π stacking interactions, intercalation-based pathway and improve the solubility in water. Herein we report the synthesis of A-ring pyridazine etoposide **13** from revisiting our synthetic approach based on the Stille coupling reaction.

The synthesis of pyridazine etoposide **13** began with bistriflate **14** obtained previously¹⁸ in four steps from etoposide **1** (Scheme 1). After screening of the reaction conditions of the Stille reaction¹⁹ in order to avoid the

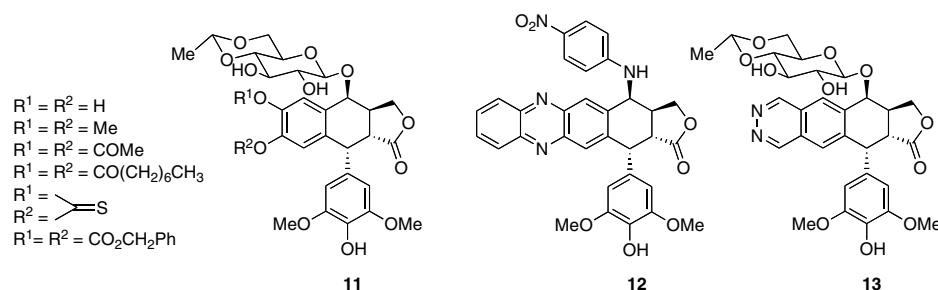
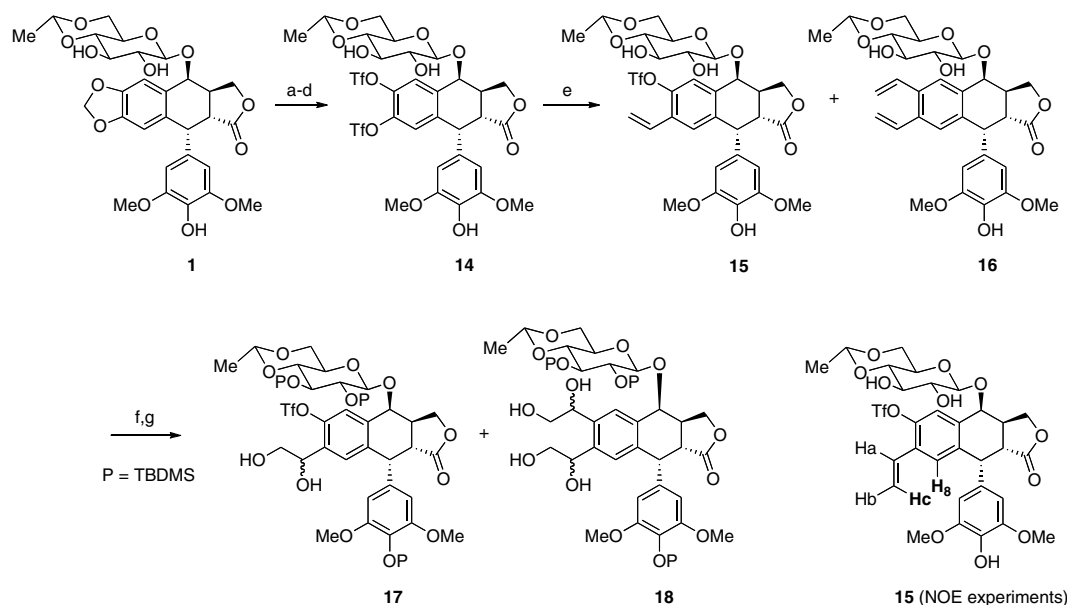
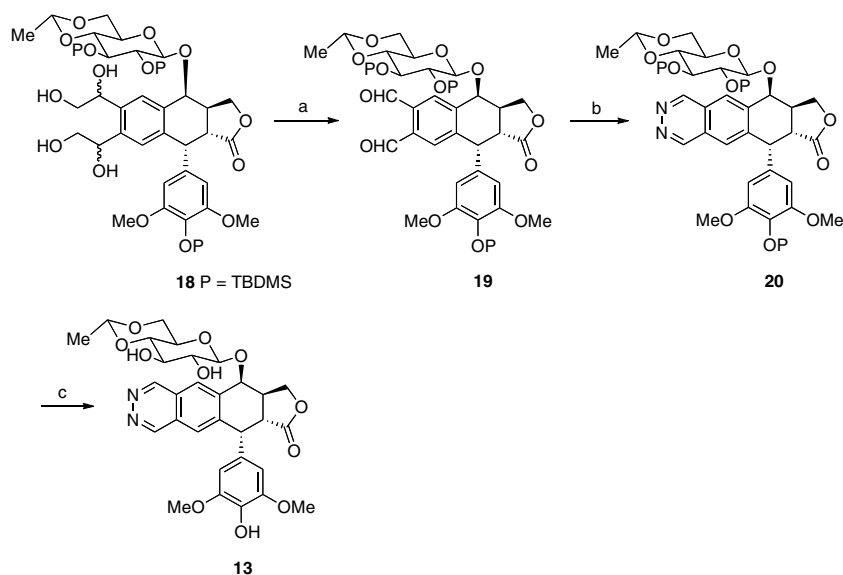


Figure 2. A-ring etoposide analogues.



Scheme 1. Reagents and conditions: (a–d) See Ref. 18; (e) Pd(PPh₃)₄ (0.2 equiv), Bu₃Sn(C₂H₃) (10 equiv), LiCl (6 equiv), dioxane, 100 °C, 1.5 h, **15** (18%) and **16** (48%); (f) TBDMSOTf, 2,6-lutidine, CH₂Cl₂, 0 °C to rt, 88% yield of triprotected monovinyl relative to **15** and 88% yield of triprotected bisvinyl relative to **16**; (g) OsO₄ cat., NMO, acetone:H₂O (8/1), rt, 5 h.



Scheme 2. Reagents and conditions: (a) $\text{Pb}(\text{OAc})_4$, benzene, rt, 2 h, 69%; (b) $\text{NH}_2\text{NH}_2 \cdot \text{H}_2\text{O}$, CH_2Cl_2 :EtOH (1/1), -55°C for 1 h, 84%; (c) $\text{HF} \cdot \text{NEt}_3$, acetonitrile, rt, 6 days, 84%.

undesired epimerization of the D-ring¹⁸ deleterious to the cytotoxic activity of this class of compounds,^{1d,e} we found that $\text{Pd}(\text{PPh}_3)_4$, under optimized conditions, gave a mixture of *trans*-monovinyl **15** and *trans*-bisvinyl **16** in a 27:73 ratio (18% and 48% yields, respectively, based on NMR analysis), which were very difficult to separate. The C-7 site of monovinylation has been determined on the basis of NOE experiments: irradiation of the H_c signal caused the enhancement of H_8 and H_b signals. Complete silylation of this mixture as TBDMS ethers followed by osmylation led to a separable mixture of diols **17** (83%) and tetrols **18** (92%).²⁰

Oxidation of this latter with $\text{Pb}(\text{OAc})_4$ gave dialdehyde **19** (2,3-*trans* relationship: $J_{2,3} = 14.1$ Hz). Formation of the A-ring pyridazine occurred when **19** was treated with hydrazine hydrate. Since, as expected, the basic TBAF led to pyridazine picroetoposide¹⁸ (TBAF, CH_2Cl_2 :THF (10/2), rt, 24 h, 72%), deprotection of **20** was carried out using $\text{HF} \cdot \text{NEt}_3$ ²¹ to furnish the target pyridazine etoposide **13**²² (Scheme 2). The TBDMS protecting group was critical for obtaining **13**. Unlike the synthesis of pyridazine picroetoposide,¹⁸ protection with the 2,2,2-trichloroethyl chloroformate group failed to cleanly deliver the key intermediate dialdehyde **19**. Synthesis of **13** using phenoxyacetyl chloride for the triprotection or benzyl chloroformate for the monoprotection at C-4' was also unsuccessful.

Pyridazine etoposide is a weak inhibitor of topoisomerase II (18% of inhibition at 50 μM for **13** versus 50% of inhibition at 50 μM for etoposide **1**). It was found inactive in vitro against the A549 human lung carcinoma epithelial cell line ($\text{IC}_{50} > 10^{-4}$ M for **13** versus $\text{IC}_{50} = 4.4$ μM for etoposide **1**), and in vivo against P388 lymphocytic leukemia in mice (no activity at a dose of 40 mg/kg for **13** versus T/C 189% at a dose of 50 mg/kg for etoposide **1**).

In conclusion, these results confirm the importance of the methylenedioxy A-ring of etoposide for optimal activity. The use of the pivotal intermediate **19** should give access to novel analogues of **1**, in particular those which have, fused to C-ring, either three rings (e.g., phthalazine) as in the case of **12** or methylenedioxy-phenyl-containing four rings, to pursue the investigation of the unexplored intercalating domain of McDonald's composite pharmacophore model.

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